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Lavoisier. Antoine Lavoisier's great accomplishments include the discovery of oxygen's role in combustion, helping to develop the metric system, writing the first extensive list of elements, helping to reform the nomenclature of chemistry, and the discovery that while matter may change shape through chemical reaction its mass remains the

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same.

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Lavoisier helped create the metric
system, wrote the first detailed list of
elements, and facilitated to reform
chemical terminology. He anticipated
the presence of silicon (1787) and was

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also the first to recognize sulfur as an
element (1777) rather than a compound.

On Lavoisier, Fourier, Faraday and the ambition of science

Lavoisier on Elements and Oxidation In
setting chemistry on the proper track for
fruitful development, one of Lavoisier's
main contributions was to systematize

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the idea of elements at a time when
some scientists were still struggling to
understand substances in terms of their
content of Air, Earth, Fire, and Water.
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**Lavoisier on Elements - Yale
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The names in the right purple column
were listed as elements but Lavoisier

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suspected that they weren't elements;
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he just couldn't get them to decompose.
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Later, lime was shown to be calcium
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combined with oxygen. Magnesia
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consisted of magnesium, sulfur and
oxygen. Silex and Argill were also found
not to be elements but made up of true
elements.

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**Antoine-Laurent Lavoisier -
PhilosophyProfessor.com**

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Many natural philosophers still viewed the four elements of Greek natural philosophy—earth, air, fire, and water—as the primary substances of all matter. Chemists like Lavoisier focused their attention upon analyzing “mixts” (i.e., compounds), such as the salts formed when acids combine with alkalis.

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It is generally accepted that Lavoisier's great accomplishments in chemistry stem largely from his changing the science from a qualitative to a quantitative one. Lavoisier is most noted for his discovery of the role oxygen plays in combustion. He recognized and

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named oxygen (1778) and hydrogen
(1783), and opposed the phlogiston
theory.

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